Theory

## Electrolysis in Organic Synthesis - Answer Sheet

| $6 \%$ of total |  |  |  |  |  |  |  |  |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| Question | 9.1 | 9.2 | 9.3 | 9.4 | 9.5 | 9.6 | 9.7 | Total |
| Points | 3 | 3 | 2 | 5 | 5 | 2 | 9 | 29 |
| Score |  |  |  |  |  |  |  |  |

9.1 (3 pt)

Provide the structural formulae of $\mathbf{A}, \mathbf{B}$, and $\mathbf{C}$.


## 9.2 (3 pt)

Formulate the oxidative and reductive half reactions and the full redox reaction.
Reductive half reaction:

Oxidative half reaction:

Full redox reaction:

Theory

9.3 (2 pt)

Provide the intermediates in the mechanism for the oxidative decarboxylation and formation of the product.


Translation:
1:
2:

Theory
9.4 (5 pt)

Provide the structures of D-H.
$\square$


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9.5 ( 5 pt )

Provide the structures for both products $\mathbf{K}$ and $\mathbf{L}$. Indicate how the two products are related.

| $\mathbf{K}$ |  |
| :--- | :--- |
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|  |  |
|  |  |


$\square$ Epimer
$\square$ Diastereomer
$\square$ Enantiomer
$\square$ Constitutional Isomer
9.6 (2 pt)

Provide the structure of $\mathbf{N}$ and $\mathbf{O}$.


Theory
9.7 (9 pt)

Provide the structural formulae of compounds $\mathbf{P}, \mathbf{Q}, \mathbf{R}$, and $\mathbf{S}$. Indicating stereochemistry is not required. Hint: $\mathbf{S}$ is a tricyclic product.


